Note

Asymmetric Synthesis of Neolignans (-**)-***epi***-Conocarpan and (**+**)-Conocarpan via Rh(II)-Catalyzed C**-**H Insertion Process and Revision of the Absolute Configuration of (**-**)-***epi***-Conocarpan**

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*Recei*V*ed March 6, 2009*

Catalytic asymmetric synthesis of neolignan natural products (-)-*epi*-conocarpan and (+)-conocarpan has been achieved by exploiting an enantio- and diastereoselective intramolecular C-H insertion reaction to construct a *cis*-2-aryl-2,3 dihydrobenzofuran ring system as a key step. The C-^H insertion reaction of 5-bromoaryldiazoacetate catalyzed by Rh2(*S*-PTTEA)4, a new dirhodium(II) carboxylate complex that incorporates *N*-phthaloyl-(*S*)-triethylalaninate as chiral bridging ligands, provided 2-aryl-5-bromo-3-methoxycarbonyl-2,3-dihydrobenzofuran with exceptionally high diastereoselectivity (*cis/trans* = 97:3) and high enantioselectivity for the *cis* isomer (84% ee).

Neolignan natural products containing a 2-aryl-2,3-dihydrobenzofuran ring, such as conocarpan, epi -conocarpan, 2^3 , 4di-*O*-methylcedrusin,³ and brugnanin,⁴ display a wide range of biological activities.⁵ Consequently, much effort has been

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devoted to the stereocontrolled synthesis of 2-aryl-2,3-dihydrobenzofuran derivatives.6

(+)-Conocarpan (**1**), first isolated from the wood of *Conocarpus erectus* by Hayashi and Thomson in 1975,¹ exhibits a diverse array of biological activities, including insecticidal,⁷ antifungal,⁸ and antitrypanosomal properties.⁹ (-)-*epi*-Conocarpan (**2**) was isolated from the roots of *Piper regnellii* by Kato and co-workers in 1999.² The absolute configuration of $(+)$ conocarpan was assigned as (2*R*,3*R*) by comparing the CD curve of conocarpan acetate with those of reference compounds possessing different substitution patterns in the aromatic rings.^{1,10} However, a later chiroptical study of 2,3-dihydrobenzofurans fused to the steroid $nucleus¹¹$ indicated that its absolute configuration should be reversed. Recently, Clive and Stoffman established the absolute configuration of $(+)$ -conocarpan as $(25,35)$ through their asymmetric synthesis of $(-)$ -conocarpan (vide infra).¹² Since the absolute configuration of $(-)$ -*epi*conocarpan was determined to be $(2*S*,3*R*)$ by CD spectra,² this assignment should also be reversed.

Racemic conocarpan can be easily synthesized either by an oxidative dimerization of p -propenylphenol with $\text{FeCl}_3^{1,13}$ or by a $Mn(OAc)$ ₃-based oxidative cycloaddition of 4-(3-chloropropyl)-2-cyclohexenone with anethole.¹⁴ Kato and co-workers reported an enantioselective conversion of *p*-propenylphenol to (+)-conocarpan by an enzyme fraction obtained from *Piper regnellii* leaves.¹³ Clive and Stoffman have accomplished an asymmetric synthesis of $(-)$ -conocarpan by means of a Sharpless asymmetric epoxidation and a radical cyclization of optically active benzylic ethers.12 The Zheng and Che groups achieved the synthesis of (\pm) -*epi*-conocarpan by employing a ruthenium(II) porphyrin-catalyzed intramolecular C-H insertion of aryl tosylhydrazone salt as a key step.¹⁵ However, to the best of our knowledge, an asymmetric synthesis of $(-)$ -epi-conocarpan has not been reported. In this respect, we have provided a protocol for asymmetric synthesis of *cis*-2-aryl-3-methoxy-

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FIGURE 1. Structure of chiral dirhodium(II) complexes.

carbonyl-2,3-dihydrobenzofurans **6** via enantio- and diastereoselective C-H insertions of phenyldiazoacetates **⁵** catalyzed by dirhodium tetrakis[*N*-phthaloyl-(*S*)-tert-leucinate], Rh₂(*S*-PTTL)₄ (**3a**) (Figure 1), wherein perfect *cis* selectivity and high enantioselectivity (up to 94% ee) are achieved (eq 1).^{16,17} In

order to demonstrate the synthetic potential of this catalytic methodology, we herein report asymmetric synthesis of $(-)$ *epi*-conocarpan and (+)-conocarpan and a revision of the absolute configuration of natural $(-)$ -*epi*-conocarpan.

SCHEME 1. Retrosynthetic Analysis of 1 and 2

Our retrosynthetic analysis of **1** and **2** is outlined in Scheme 1. It has been documented that thermodynamically less-stable *cis*-2-(4-hydroxyphenyl)-2,3-dihydrobezofurans are epimerized to *trans* isomers by treatment with an acid (TFA in CH_2Cl_2)¹⁸ or a base (Na₂CO₃ in MeOH).¹⁹ Thus, synthesis of 1 would be achieved by epimerization at the C2 stereocenter of **2**. We carefully selected the TIPS ether as the protecting group for the phenolic moiety, reasoning that a TIPS-protected *p*-cresol is stable under basic conditions.²⁰ It was anticipated that a propenyl group at C5 would be introduced by a Suzuki-Miyaura coupling of 5-bromo-2,3-dihydrobenzofuran **8a** with propenylboronic acid (7) .²¹ As mentioned above,¹⁶ we envisioned that Rh2(*S*-PTTL)4-catalyzed C-H insertion of methyl [5-bromo-2-(4-triisopropylsilyloxybenzyloxy)phenyl]diazoacetate (**9a**) would provide *cis*-(2*R*,3*S*)-5-bromo-2,3-dihydrobenzofuran **8a**. C-^H insertion reactions of 5-bromoaryldiazoacetates catalyzed by chiral dirhodium(II) complexes have already been reported by two research groups (eq 2). Fukuyama and co-workers dem-

SCHEME 2. Preparation of 5-Bromoaryldiazoacetate 9a

mixtures of *cis* and *trans* isomers with 32% ee for **10b**. 18,22 Davies and co-workers reported that $Rh_2(S-PTAD)_4$ (3c),²³ derived from adamantylglycine, is an effective catalyst for the reaction of **9b**, providing stereoisomers **8b**/**10b** in 72% yield with a 14:1 preference for the *cis* isomer **8b** and 79% ee for **8b**. 23a,24

5-Bromoaryldiazoacetate **9a** was prepared as shown in Scheme 2. *O*-Alkylation of methyl (5-bromo-2-hydroxyphenyl) acetate (**11**) ²⁵ with 4-(triisopropylsilyloxy)benzyl bromide (**12**) 26 provided 5-bromoarylacetate **13** in 68% yield. Since attempted direct diazo transfer to **13** with *p*-acetamidebenzenesulfonyl azide and DBU in $CH₃CN$ gave decomposition products, we used the Danheiser modification of Regitz's diazo transfer reaction.²⁷ Deprotonation of 13 with LiHMDS in THF at -78 °C followed by treatment of the resulting enolate with trifluoroethyl trifluoroacetate afforded a trifluoroacetylated product, which upon diazo transfer with methanesulfonyl azide and triethylamine in CH3CN gave 5-bromoaryldiazoacetate **9a** in 75% yield.

On the basis of our previous work,¹⁶ we initially explored the intramolecular C-H insertion of **9a** in toluene using 1 mol

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^a All reactions were carried out as follows: Rh(II) catalyst (1 mol %) was added to a solution of **9a** (0.10 mmol) in the indicated solvent (1 mL) at the indicated temperature. *^b* Combined yield of *cis* and *trans* isomers. *^c* Determined by ¹ H NMR analysis of the crude product. *^d* Determined by HPLC (Chiralcel OD-H). *^e* Determined by HPLC (Chiralpak IA). *^f* Isolated yield of *cis* isomer.

% of $Rh_2(S-PTTL)_4$ (3a) at -23 °C (Table 1, entry 1). The reaction proceeded to completion in less than 2.5 h, giving a 98:2 *cis*/*trans* mixture of 2-aryl-5-bromo-2,3-dihydrobenzofurans **8a**/**10a** in 62% combined yield. The enantiomeric excess of the *cis* isomer **8a** was determined to be 53% by HPLC analysis (Chiralcel OD-H). The preferred absolute configuration of **8a** was assigned as (2*R*,3*S*) by single-crystal X-ray analysis (see Supporting Information).28,29 Switching the solvent from toluene to $CH₂Cl₂$ improved the reaction rate, product yield and enantioselectivity, although a considerable drop in diastereoselectivity was observed $(cis/trans = 89:11$, entry 2). While lowering the reaction temperature to -40 °C had little impact on selectivities (*cis/trans* $= 84:16, 76\%$ ee, entry 3), the reaction at -60 °C increased enantioselectivity but significantly reduced diastereoselectivity (85% ee, $cis/trans = 71:29$, entry 4). The addition of 4 Å molecular sieves (MS) at -60 °C led to minor improvements in product yield and diastereoselectivity with a slight erosion in enantioselectivity (78% yield, c *is/trans* $= 74$: 26, 82% ee, entry 5).

To further enhance the diastereoselectivity, we directed our efforts to the development of new dirhodium(II) carboxylate catalysts. Eventually, we found that the reaction with dirhodium(II) tetrakis[N-phthaloyl-(S)-triethylalaninate], Rh₂(S-PT-TEA)4 (**3b**) (vide infra), characterized by an exceptionally large triethylmethyl group, proceeded at -40 °C to afford 2,3dihydrobenzofurans **8a**/**10a** in 87% yield as a 98:2 mixture of stereoisomers with 82% ee for **8a** (entry 6). Although the reason for the decreased diastereoselectivity in the reaction using Rh₂(*S*- $PTTL$)₄ (3a) at a lower temperature is unclear at present, $Rh_2(S-$ PTTEA)4 (**3b**) proved to be more effective than **3a** in terms of diastereoselectivity. Aside from high *cis* selectivity, the highest enantioselectivity (89% ee) was observed at -60 °C, although the product yield was only 34% (entry 7). Gratifyingly, this problem could be overcome with the addition of 4 Å MS as a desiccant. The reaction in the presence of 4 Å MS afforded *cis* isomer **8a** in 80% yield with a similar level of enantioselectivity (84% ee, entry 8).

The new dirhodium(II) carboxylate catalyst **3b** was prepared from the known carboxylic acid **14**³⁰ (Scheme 3). Bromination of 14 with bromine and PCl_3^{31} followed by azidation with NaN_3 provided α -azidocarboxylic acid 15. Optical resolution of 15 was accomplished through formation of diastereomeric carboximides with (S) -4-benzyl-2-oxazolidinone³² as a resolving reagent. The absolute configuration of more polar isomer **16** was established as $(4S,2'S)$ by single-crystal X-ray analysis.²⁸ Hydrolytic removal of the chiral auxiliary 33 and azide reduction followed by recrystallization from MeOH-H2O furnished amino acid **18**. $Rh_2(S\text{-PTTEA})_4$ (3b) was prepared from amino acid **18** in 72% yield according to the reported procedure for the preparation of Rh₂(*S*-PTTL)₄ (3a).³⁴

With the efficient construction of (2*R*,3*S*)-*cis*-2-aryl-5-bromo-2,3-dihydrobenzofuran **8a** realized, the stage was now set for the completion of the asymmetric synthesis of $(-)$ -*epi*-cono-

⁽²⁸⁾ CCDC 703985 (**8a**) and CCDC 695667 (**16**) contain the crystallographic data for this paper. These can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

⁽²⁹⁾ The preferred absolute configuration of *trans* isomer **10a** was determined to be (2*S*,3*S*) by comparison of the sign of optical rotation with that of the compound obtained by epimerization of **8a** at the C3 stereocenter. See Supporting Information.

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carpan (**2**) and (+)-conocarpan (**1**) as illustrated in Scheme 4. Reduction of (2*R*,3*S*)-**8a** (84% ee) with DIBAL-H was followed by tosylation of the resultant hydroxy group to afford tosylate **²¹** in 93% yield. Trituration of **²¹** [84% ee, mp 69.0-71.0 °C, $[\alpha]_{\text{D}}^{21}$ +4.9 (*c* 1.18, CHCl₃)] in hexane-EtOAc produced
optically pure material [mp 54.0–55.0 °C [α]²²₂ +6.2 (*c* 1.15 optically pure material [mp 54.0–55.0 $\rm{^{\circ}C}$, $\rm{[}\alpha\rm{]}^{22}$ _D +6.2 (*c* 1.15, CHCl³¹) from the mother liquor in 83% vield. Suzuki–Miyaura CHCl3)] from the mother liquor in 83% yield. Suzuki-Miyaura coupling of **21** (>99% ee) with propenylboronic acid **7** in the presence of $Pd(PPh_3)_2Cl_2$ and K_2CO_3 provided the coupling product 22 in 76% yield. Reduction of 22 with LiEt₃BH followed by removal of the TIPS protecting group with TBAF in THF-AcOH furnished $(-)$ -*epi*-conocarpan (2) in 83% yield without epimerization of the C2 stereocenter. The spectroscopic data (1 H and 13 C NMR, IR, and CD) of this product were consistent with those reported for natural $(-)$ -2.² Although an optical rotation of synthetic product (2R 3S)-2. $\lceil \alpha \rceil^2 \rceil_2$ -9.7 optical rotation of synthetic product $(2R,3S)$ -**2**, $[\alpha]^{21}$ _D -9.7
(c 0.45 MeOH) was lower than that of natural (-)-**2** Iit^2 $[\alpha]^{21}$ _D $(c$ 0.45, MeOH), was lower than that of natural $(-)-2$ [lit.² [α]²¹_D -33 3 (c 0.03 MeOH)], the enantiomeric purity of synthetic -33.3 (*^c* 0.03, MeOH)], the enantiomeric purity of synthetic $(-)$ -2 was confirmed to be $>99\%$ by HPLC analysis (Chiralcel OD-H). Based on the identical CD spectrum and sign of optical rotation of synthetic (2*R*,3*S*)-**2** compared to those reported for natural $(-)$ -*epi*-conocarpan,² the absolute configuration of $(-)$ *epi*-conocarpan should be revised to be (2*R*,3*S*). Epimerization of $(-)$ -2 with Na₂CO₃ in MeOH^{19,35} followed by recrystallization from hexane-EtOAc provided (+)-conocarpan (**1**) in 71% yield. Synthetic material 1 was spectroscopically (¹H and ¹³C NMR, and IR) identical to natural $(+)$ - 1^1 and also had an optical rotation, $[\alpha]^{22}$ _D +117 (*c* 1.18, MeOH), in good agreement
with the literature value $[\text{lit}^{10}$ $[\alpha]^{21}$ _D +122 (*c* 1.03, MeOH) with the literature value $\left[\text{lit.}^{10} \left[\text{Cl}^{21} \text{D} +122 \text{ (c 1.03, MeOH)} \right] \right]$.
In summary, the Rh₂(S-PTTEA)-catalyzed intramolecular In summary, the Rh₂(*S*-PTTEA)₄-catalyzed intramolecular ^C-H insertion reaction of 5-bromoaryldiazoacetate showed considerable promise for asymmetric synthesis of *cis*-2-aryl-5 bromo-2,3-dihydrobenzofurans. While Rh₂(*S*-PTTL)₄ has proven to be optimal in a wide range of enantioselective C-H insertion reactions,^{16,36} newly developed Rh₂(*S*-PTTEA)₄ has been found to exhibit even higher diastereoselectivity $(cis/trans = 97:3)$ and slightly higher enantioselectivity (84% ee) compared to those of Rh₂(*S*-PTTL)₄ in this system. Using this catalytic methodology, we have achieved the first asymmetric synthesis of $(-)$ -*epi*-conocarpan (2) in nine steps and 20% overall yield from methyl arylacetate 11 and an asymmetric synthesis of $(+)$ conocarpan (1) via epimerization of $(-)$ -2. We also described a revision of the originally reported absolute configuration of (-)-*epi*-conocarpan. Further application of this methodology to asymmetric synthesis of biologically active neolignans containing a 2,3-dihydrobenzofuran skeleton is currently in progress.

Experimental Section

Representative Procedure for the C-**H Insertion of 9a (Table 1, entry 8).** $Rh_2(S\text{-PTTEA})_4$ **(3b)** \cdot 2EtOAc (47.7 mg, 0.03) mmol, 1 mol %) was added to a mixture of **9a** (1.60 g, 3.0 mmol) and 4 Å MS (1.60 g) in CH₂Cl₂ (30 mL) at -60 °C under Ar atmosphere. After stirring at this temperature for 24 h, the 4 Å MS was filtrated through a Celite pad, and the filtrate was concentrated in vacuo. The ratio of **8a/10a** was determined to be 97:3 by ¹H NMR of the crude product. The residue (1.7 g) was purified by column chromatography (silica gel 90 g, $25:1$ hexane/Et₂O) to give **8a** (1.21 g, 80%) as a white solid: $R_f = 0.50$ (6:1 hexane/Et₂O); mp 62.0–63.0 °C for 84% ee; $[\alpha]_{D}^{21}$ – 31.4 (*c* 1.15, CHCl₃) for 84% ee; IR (CHCl₃) ν 1738 cm^{-1, 1}H NMR (270 MHz, CDCl₃) δ 84% ee; IR (CHCl₃) *ν* 1738 cm⁻¹; ¹H NMR (270 MHz, CDCl₃) δ 1.08 (d, J = 7.2 Hz, 18H), 1.20–1.29 (m, 3H), 3.27 (s, 3H), 4.56 $(d, J = 9.9 \text{ Hz}, 1\text{H}), 5.94 (d, J = 9.9 \text{ Hz}, 1\text{H}), 6.80-6.85 \text{ (m, 3H)},$ 7.17 (d, *J* = 8.6 Hz, 2H), 7.33-7.36 (m, 2H); ¹³C NMR (100 MHz, CDCl3) *δ* 12.6, 17.9, 51.8, 53.6, 86.1, 111.4, 112.8, 119.7, 127.0, 127.5, 128.9, 129.0, 132.3, 156.3, 159.6, 169.7; HRMS (EI) calcd for $C_{25}H_{33}O_4SiBr(M^+)$ 504.1331, found 504.1332. Anal. Calcd for C25H33O4SiBr: C, 59.40; H, 6.58; Br, 15.81. Found: C, 59.11; H, 6.36; Br, 16.11. The enantiomeric excess of **8a** was determined to be 84% by HPLC with a Chiralcel OD-H column (100:1 hexane/ *i*-PrOH, 0.5 mL/min): t_R (major) = 20.6 min for (2*R*,3*S*)-8a; t_R $(\text{minor}) = 25.6 \text{ min}$ for $(2*S*,3*R*)$ -8a.

Acknowledgment. This research was supported, in part, by a Grant-in-Aid for Scientific Research on Priority Areas "Advanced Molecular Transformations of Carbon Resources" from the Ministry of Education, Culture, Sports, Science and Technology, Japan. We thank S. Oka, M. Kiuchi, and T. Hirose of the Center for Instrumental Analysis at Hokkaido University for mass measurements and elemental analysis.

Supporting Information Available: Full experimental, characterization data, and copies of ${}^{1}H$ and ${}^{13}C$ NMR spectra for all new compounds, as well as X-ray crystallographic data in CIF format for **8a** and **16**. This material is available free of charge via the Internet at http://pubs.acs.org.

JO900502D

⁽³⁵⁾ The epimerization afforded a 12:1 mixture of $(+)$ -conocarpan (1)/ $(-)$ -*epi*-conocarpan (2) in 98% yield. These isomers could not be separated by column chromatography on silica gel.

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